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Journal article

Reduction and mitigation strategy of carbon dioxide emissions from internal combustion engine: An engine development initiative for sustainable environment

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1 **Reduction and Mitigation strategy of Carbon Dioxide Emissions**
2 **from Internal Combustion Engine: An Engine Development**
3 **Initiative for Sustainable Environment**

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22 **Research Highlights**

- 23 • Methanol-Gasoline blending to test and compare CO₂ emissions
24 • Engine testing for emissions and performance
25 • Mitigation of CO₂ emissions through a four muffler design replacement
26 • Sustainability study of replacing the fuel and exhaust muffler in CO₂ reduction.

27 **Abstract:**

28 This study reports various methods to reduce and mitigate the CO₂ emissions from internal
29 combustion engines. In order to achieve this, the gasoline fuel is replaced with Gasoline-
30 Methanol blend of 5%, 10% and 15% of methanol by volume. Then the emissions and
31 performance tests are carried out to catch CO₂ emissions from a spark ignition engine operating
32 in different combinations of torque (2Nm, 3Nm and 5 Nm) and speed (500 rpm, 1000 rpm and

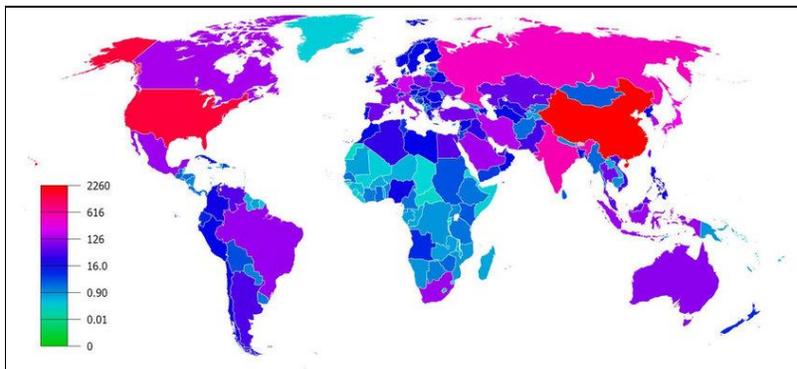
33 1500 rpm). In order to further mitigate the CO₂ emission, four different design mufflers are
34 manufactured and replaced for testing to see the effect on CO₂ emissions.

35 **Keywords:** CO₂; mitigation; methanol blend; muffler; chamber; turbo type; gasoline

36 **1. Introduction**

37 Carbon dioxide (CO₂) emissions are one of the major constituents of greenhouse gases.
38 Automobiles are a major source of CO₂. Rapid increase in the use of automobiles, powered by
39 fossil fuels, emit CO₂, CO, NO_x and HC on a large scale to the environment. Urban areas are
40 more prone to environmental degradation compared to rural ones due to increased vehicle
41 density. This creates an imbalance of atmospheric constituents, leading to a health hazard to
42 people. Mega cities like Delhi, Shanghai, Tokyo etc. have already passed through the phase of
43 the dangerous effects of CO₂ emissions. Society is so dependent on vehicles that it is difficult to
44 consider an alternative way of moving around. Alternative power systems that produce less CO₂
45 emissions, such as electric vehicles and hybrid technologies for vehicles are being developed, but
46 their current cost and the supportive infrastructure is prohibitive for most cities. Indeed, even
47 well-developed economies are introducing these at a slow and steady pace to enable technology
48 to be established (ref). In the UK, the overall carbon target to largely decarbonized road transport
49 sector by 2050. To achieve this from 2040 onwards every single car or van to be sold to be ultra-
50 low emission vehicles ([Office of low carbon vehicle UK, 2013](#)). For many more years the
51 internal combustion engine will remain an inherent part of vehicles in most regions in the world.
52 The current requirement is to reduce the emission levels without significantly modifying the
53 engine infrastructure. Many attempts are being made in to reduce emission from engines
54 operated on fossil fuel. One of the promising methods is to replace traditional fuel (gasoline/
55 diesel) with fuels which have been blended with lighter components. The objective of this work

56 is to adopt the fuel variation in the engine through replacements of Gasoline with B₅, B₁₀ and
57 B₁₅. Also to implement muffler design modification for monitoring the trend of CO₂ in an SI
58 engine.



59

60 **Fig.1** World Carbon dioxide emission levels (this figure to be updated)

61 **2. Background Motivation**

62 Carbon dioxide (CO₂) from automotive vehicles is a major source of emissions. It is a
63 contributor to climate change. Increases in atmospheric temperatures and rising sea levels are the
64 indicators of a heavy presence of CO₂ in the atmosphere (Ekwurzel et al., 2017). (Terrenoire et
65 al., 2007) carried out anthropogenic CO₂ emission recording for 343 cities. Here data from
66 individual cities are subject to quality control to separate from those of other greenhouse gases.
67 Through this analysis, some set of ancillary data from other sources (socio-economic and traffic
68 indices) or calculated (climate indices, urban area expansion) and combined with emission data.
69 (Aye et al., 2017) studied the effect of economic growth (EG) on CO₂ emission using a dynamic
70 panel threshold framework. The results show the EG has a negligible effect on CO₂ emissions.
71 There is evidence of a significant causal relationship between CO₂ emission, economic growth,
72 energy consumption and financial development. The findings emphasize the need for

73 transformation of low carbon technologies aimed at reducing emissions and sustainable
74 economic growth. This may include energy efficiency and switching away from non-renewable
75 energy to renewable energy. (Abeydeer et al., 2019) identified sulphur dioxide, nitrogen dioxide
76 and carbon dioxide as prime causes of global climate change. Out of them, CO₂ is recognized as
77 a good agent for exploring the strategy for carbon reduction and mitigation. It can be seen that
78 evaluating greenhouse gas emissions and estimating the carbon footprint is a preferred method
79 for most research in this area. Moreover, climate change and environmental effects of carbon
80 emissions were also significant points of concern in carbon emission research. The key findings
81 of this study will be beneficial for the policymakers, academics, and institutions to determine the
82 future research directions as well as to identify with whom they can consult to assist in
83 developing carbon emission control policies and future carbon reduction targets. (Valihesari et
84 al., 2019) tested a blend of gasoline, oxygenate additive; Methanol and metal nanoparticles:
85 Fe₂O₃ and TiO₂ in a 4-stroke engine to investigate the effects of the new blend on the engine
86 parameters, such as power and torque and also the amount of target pollutant gases emitted
87 which are CO₂, CO, NO_x and HC. The research being undertaken, around the world is now
88 mainly focusing on reducing the engine emissions, while using fossil fuels. (Verhelst et al.,
89 2019) carried out a review on the use of methanol as a pure fuel or blend component for ICES.
90 They summarized various method of methanol production and also the health and safety issues
91 associated with the use of methanol as fuel for ICES. Many properties of methanol (for example
92 high heat of vaporization) are superior to that of gasoline. These help make blended fuels a
93 suitable improved alternative to traditional automotive fuel compositions. It is necessary to
94 address changes in hardware, materials and heat recovery to improve the engine efficiency when
95 using methanol. Furthermore, the behaviour of methanol fuel such as, mixture formation,
96 normal/abnormal combustion, high latent heat, fast burning velocity, high knock resistance etc.

97 are reviewed for the modelling aspect. Blended fuels show promising performance as compared
98 to traditional gasoline or diesel fuel. (Shrivastava et al., 2019) through transesterification of a
99 bio-diesel from karanja and roselle oil, tested for emissions and performance. They achieved
100 lower thermal efficiency with reduction of exhaust gas temperature by 1.48% and 1,38%. But
101 brake specific fuel consumption increased by 4.13% compared to traditional diesel fuel. The use
102 of blend shows 15.3% less NO_x and 1.92% more CO₂ compared to diesel. Through this analysis,
103 an ANN model was developed to predict the output parameter through multi variable response.
104 (Mourad et al., 2019) studied the blending of Gasoline-ethanol and Gasoline-butanol
105 (25,5%,10%,15%) on the emissions and power of an engine. They observed a 13.7% reduction in
106 CO₂, 25.2% reduction in hydrocarbon, 8.22% reduction in fuel consumption. However they also
107 reported an 11.1% reduction in engine power.

108 Due to the uncontrolled use of vehicles operating on fossil fuels, regulations are defined to curb
109 emission level at a regional or country wide level. (Olabi et al., 2020) carried out a review of the
110 regulations and techniques to eliminate toxic emissions from diesel engine cars. (Rao et al.,
111 2018) carried out a review on Performance of the IC Engine Using Alternative Fuels. An attempt
112 is made to design and develop IC engine parts that are most suitable for alternate fuels, that can
113 last longer without affecting the performance of the engine.

114 (Mishra et al., 2020; Gupta et al., 2019) aimed to fully evaluate the effects of petrol-methanol
115 blends on the emission and performance of engines and the corresponding noise levels. Petrol
116 blended with 5%, 10% and 15% of methanol was used in three separate tests, which are
117 conducted at constant torque and variable speed conditions. The exhaust emission analysis was
118 done using a six gases emissions analyzer. The emission levels were measured, while the engine
119 was mounted in a special purpose engine test bed fitted with an eddy current dynamometer
120 capable of controlling the speed and torque of the engine. The noise level of the silencer was also

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121 measured to understand the effects of methanol percentage on engine knock. The analysis
122 predicts the blend of 5%, 10% and 15% methanol with petrol exhibited less emissions and
123 knocking behaviour compared to pure petrol. In some cases, the NO_x emissions of richer fuel
124 blends was higher than that of leaner ones. However, other emission constituents were
125 significantly reduced when using the methanol blend in place of pure petrol.

126 Based on this broad literature review, it is understood that most of the research concentrated on
127 use of blending, while few of them reported modelling and simulation. Very few reported the
128 cumulative effect of fuel blends and engine modification on emissions, especially CO₂ and its
129 reduction and mitigation strategy.

130 **2. Materials and Methods**

131 Different techniques are considered to achieve a reduction and mitigation of CO₂ in an ICE. One
132 such technique is to replace pure gasoline with a gasoline-methanol blend. It is not possible to
133 operate the engine with pure methanol as the fuel is hugely toxic and burns with an invisible
134 flame. Furthermore, pure methanol is very corrosive to the engine components (). To manage
135 these negative effects, the methanol percentage is maintained at low levels of 5%, 10% and 15%
136 respectively.

137 *2.1 Gasoline-Methanol blend preparation and characterization*

138 The composition of blend in this study is made to (95% by vol. of Gasoline and 5% by vol. of
139 Methanol) B₅, (90% by vol. of Gasoline and 10% by vol. of Methanol) B₁₀ and (85% by vol. of
140 Gasoline and 5% by vol. of Methanol) B₁₅. The blending is done manually, and is effective due
141 to the high diffusivity of methanol.

142 The density of the blend of two liquids can be numerically computed as per equation (1)

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143
$$\rho_{blend} = \frac{\rho_g V_g + \rho_m V_m}{V_g + V_m} \quad (1)$$

144 Again, kinematic viscosity for the blend can be estimated numerically using three different
 145 methods; Gambill method (Gambill et al., 1959), Refuta equation and chevron formula as given
 146 in Eq. (2), Eq(3) and Eq(5) respectively.

147
$$\kappa_{blend} = x_g \kappa_g^{\frac{1}{3}} + x_m \kappa_g^{\frac{1}{3}} \quad (2)$$

148 As per (Maples et al., 2000), Refuta-mass fraction basis
 149 yields $VBN_i = 14.534 \times \ln(\ln(\kappa_i + 0.8)) + 10.975 \quad (3)$

150
$$VBN_{blend} = \sum_{i=0}^{n=0} x_i VBN_i \quad (4)$$

151
$$\kappa_{blend} = \exp\left(\exp\left(\frac{VBN_{blend} - 10.975}{14,534}\right)\right) - 0.8 \quad (5)$$

152 Chevron formula (volumetric basis)

153
$$VBN_i = \frac{\ln(\kappa_i)}{\ln(1000 \times \kappa_i)} \quad (6)$$

154
$$VBN_{blend} = \sum_{i=0}^{n=0} v_i VBN_i \quad (7)$$

155 The octane number of a blend is calculated on basis of the formula given in Eq. (8)

156
$$OCT_{blend} = OCT_{gasoline} \times (v_{gasoline}) + OCT_{methanol} \times (v_{methanol}) \quad (8)$$

157 Once the blend is ready, the fuel characteristics of the blend are studied with particular focus on
 158 how those desired properties are similar to the pure gasoline which is being replaced. Table 1
 159 shows the comparative value of B₅, B₁₀ and B₁₅ with gasoline and methanol.

160 Table 1 Comparative values of blend with respect to gasoline and methanol

<i>Fuel properties</i>	<i>Testing standard</i>	<i>Gasoline</i>	<i>B5</i>	<i>B10</i>	<i>B15</i>	<i>Methanol</i>
Density (kg/L)	ASTM 4052	0.745	0.742	0.736	0.732	0.784
Kinematic viscosity (C Stoke)	ASTM D445	0.88	0.86	0.84	0.81	0.65
Acid value (mg KOH/g)	ASTM D664	0.32	0.30	0.295	0.29	12.5
Flash point (°K)	ASTM D93	246	244	242	241	285
Calorific value (kJ/kg)	ASTM D240	47300	47230	47129	46965	22700
Auto-ignition temperature (°C)	ASTM E659	580	573	564	540	420
Octane number	ASTM D2700 (MON)	92	96	98	99	130
	ASTM D2699 (RON)	100	100	101	101	102

161
 162 *2.2 Material for muffler manufacturing*

163 In this study, one of the CO₂ mitigation techniques is considered to be muffler design
 164 modification. The mufflers are manufactured out of GI pipes and sheets ($E=200GPa$, $\nu=0.29$).
 165 The fabrication process includes, metal sheet forming, welding, hole drilling and assembly. All
 166 the activities are performed in the Central Workshop of KIIT University Bhubaneswar. Fig. 2
 167 shows the four different mufflers prepared for this analysis.

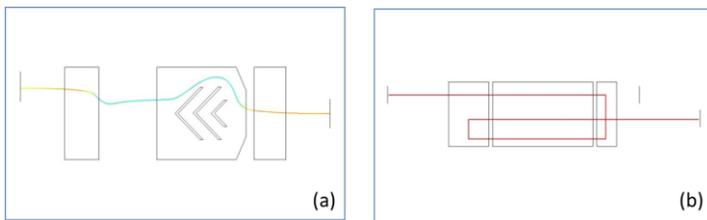
Commented [RI3]: Do you mean GCI? I don't know what is GI? Grey Iron. Would this not be very corrosive? There are many materials available that would be more suited to the environment.

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(a) Type-A (b) Type-B (c) Type-C (d) Type-D

168
 169 **Fig.2** Muffler manufacturing, (a) Type A: Chamber non-perforated, (b) Type B: Chamber perforated, (c)
 170 Type C: Turbo non-perforated and (d) Type D: turbo perforated



171
 172 **Fig.3** Exhaust gas path (Mishra et al., 2020) in: (a) Chamber type muffler, (b) Turbo type muffler

173 *2.3 Details of Engine-Test bed-Emission Analyzer*

174 The engine used for this particular testing of emissions and performance is a four stroke, single
 175 cylinder, double valve, 105.6 cc engine (Mishra et al., 2020). It has bore to stroke dimension of
 176 49.0 x 56.0 mm and a compression ratio of 9:1. It can deliver power maximum up to 6 kW with
 177 7500 rpm. It has the provision of wet sump lubrication, fin cooling and used wet type multi-plate
 178 clutch. It is fitted with a four-speed constant mesh type gear transmission system. Figure 4(d)
 179 shows the engine of this specification mounted in the test bed for testing.



180

181

Fig.4 Emission measurement, (a) Emission data acquisition, (b) performance data acquisition, (c) emission sensing at exhaust and (d) Engine test bed

182

183

The table 2 shows the specification of the dynamometer for this study. The dynamometer used here is eddy current type with maximum engine torque of 90 Nm and 7000 rpm. APPSYS WED 38S type magnetic water strainer is used here along with a water flow switch, reaction type torque sensor, torque calibration arm and magnetic pick up sensor. The control panel is equipped with PC hardware, PCI data card and a data acquisition system to view and control the torque (Nm), speed (rpm), mechanical power (kW/HP), pressure (N/m²) and temperature (°C). Data acquired can be stored in an excel sheet along with various graphical outputs. The dynamometer can be controlled either in automatic mode or manual mode.

187

188

Table 2 Specification of water-cooled Eddy current dynamometer with 38 kW power rating (Mishra et al., 2020)

Attributes	Details
Model	APPSYS WED 38S

Rated absorption Power (KW)	38 kW (50 hp).
Maximum torque (Nm)	90 Nm
Maximum Torque at Speed Range	1400 to 4031 rpm
Maximum Speed (R / Min)	7000 rpm (for Speed more than 7000 rpm high speed bearings are used.)
Torque measurement precision (F. S.)	± 0.5 FS%, 0.1 Nm resolution
Speed measurement precision (F. S.)	± 0.5 FS %, 1 rpm resolution
The direction of rotation	Both Direction, Clock wise & Anti-Clock wise
Max. Water Flow (Ltrs / hr) with Pressure	1400 Ltrs/hr at 1 – 2 bar pressure
Drainage maximum temperature (° C)	65
Moment of inertia (kgm ²)	0.018

193

194 *2.4 Emission measurement using HORIBA MEXA-584L Emission Analyzer*

195 The emission analyzer used in this study is the HORIBA MEXA-584L, which can
 196 simultaneously sense CO, HC and CO₂ using non-dispersive infrared (NDIR) technique. The air-
 197 to-fuel-ratio or excess air ratio (A) is also measured with this analyzer. The analyzer is a mobile
 198 system, it can even be used outdoors and has a single screen. Also, O₂, NO_x, engine speed and oil
 199 temperature can be measured in this instrument. Table 3 provides the detail specifications of the
 200 emission analyzer.

201 **Table 3** Specification of Horiba Mexa 584-L emission gas analyzer (Mishra et al., 2020)

<i>Attributes</i>	<i>Details</i>
Measured gas components (standard)	<ul style="list-style-type: none"> • CO, CO₂, LAMBDA (Unburnt HC), O₂ and NO_x
Measuring principle	<ul style="list-style-type: none"> • CO, HC, CO₂: Non-dispersive infrared (NDIR) • Air-to-fuel ratio (AFR), Lambda: Carbon balance method or Brettschneider method with O₂ measurement. AFR and lambda are calculated by carbon balance in standard configuration.
Conformed standard	<ul style="list-style-type: none"> • OIML Class 0-CE-FCC.
Ambient humidity	<ul style="list-style-type: none"> • Under 90% relative humidity.

202

203 The fig. 4(a) shows this emission analyzer in action, measuring exhaust gases from the engine. It
204 should be ensured that the source of power is stable. Before switching on the analyzer, it is
205 ensured that the sensing pipe end is made leak proof using a rubber cap. After a warmup period
206 of 300s, it automatically starts the leak detection test. If it fails the leak detection test, the leak
207 proofing should be inspected and the procedure is repeated. If it passes the leak detection test, the
208 measuring of HC following the removal of the cap can be undertaken. Once the HC hang up test
209 was done, the analyzer is ready to measure the emissions from the engine. There is one
210 communication software in MEXA-584L, which can interface the machine with the computer
211 with sampling rate and sampling time. The data can be recorded once every 3s for 120s and
212 stored in an excel sheet.

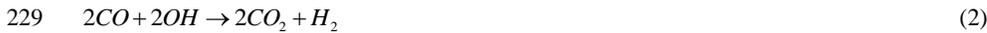
213 *2.5 CO₂ Emission formation mechanism*

214 There are basically three types of emission formed in a running engine; exhaust emissions, crank
215 case emissions and evaporative emissions. In this analysis, we shall consider the exhaust
216 emissions. As per earlier studies ([Gupta et al., 2019](#)), 100 % CO/CO₂ emissions are from the
217 exhaust of combustion gases. The emissions formation mechanism of CO₂ is a two-step process
218 ([NPTEL IITK, 2012](#)). The first step is conversion of HC to CO, where several oxidation
219 reactions are involved in the formation of intermediate compounds like small HC molecules,
220 aldehyde, ketones etc. as given in equation (1)



222 With availability of sufficient oxygen in air conversion of CO to CO₂ is ensured. The chemical
223 reaction is given in equation (2). The formation of CO₂ is the reassurance that complete
224 combustion has occurred and sufficient oxygen and time were available to eliminate unburned

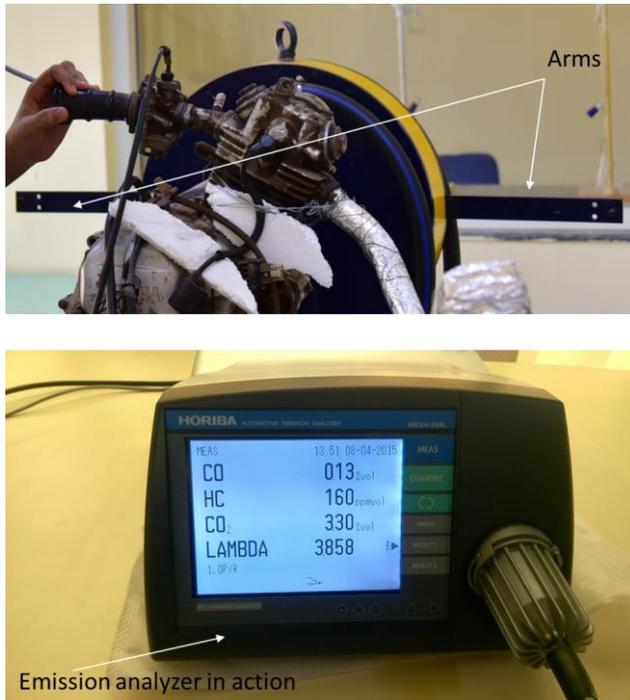
225 HC from the exhaust gases. However, the quantity of CO₂ is a directly related to the
226 performance of the engine. The trend in modern engines is to reduce fuel consumption and thus
227 reduce the CO₂ emissions by reducing fuel consumption, improving combustion processes and
228 reducing the overall engine mass and friction through improved engine refinement.



230 *2.6 CO₂ Emission Measurement*

231 In this study, we have implemented a single cylinder spark ignition engine, the details of which
232 is given in fig 4(d) earlier. The engine is operated at different combinations of torque and speed
233 to acquire the CO₂ emissions by using the HORIBA MEXA emissions analyzer. The testing is
234 done by using the different gasoline blends, B₅, B₁₀ and B₁₅ to obtain three different sets of
235 results for the fuels. A universal engine test bed fitted with an eddy current dynamometer is
236 engaged to mount the engine. Through this arrangement, this engine is mounted on a universal
237 test bed that is equipped with an eddy current dynamo meter with control panel arrangement for
238 load, torque and speed monitoring through digital/PC mode. Before recording performance
239 parameters, the dynamometer is subjected to a 'load calibration test' to ensure that the sensed
240 digital data and the computerized data are correct. A 10 kg weight is used to perform this
241 calibration. As the load arm is 50 cm, the torque monitor should show 49 Nm torque reading on
242 both left and right side of the arms as shown in fig. 5. It is done to ensure that the digital as well
243 as computerized data acquisition systems are accurate. Continuous water circulation into the
244 eddy current dynamo meter is ensured by an external pump arrangement to extract the frictional
245 heat out of the dynamometer due to engine braking. Such monitoring is done by observing the
246 green color of the indicator light provided in the data acquisition monitor for dynamometer water
247 supply.

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Would be clearer to understand.



248

249

Fig.5 Dynamometer arm and emission analyzer in measuring mode

250

3. Results and Discussions

251

3.1 Emission Analysis Results and Discussion

252

Fig. 6(a-d) show CO₂ response to engine torque (2Nm, 3 Nm, 5 Nm) at 500 rpm. In all cases

253

(Gasoline, B₅, B₁₀ and B₁₅) chambered type muffler (Type-A and Type-B) shows less CO₂

254

emission compared to turbo type muffler (Type-C and Type-D). Chambered type non-perforated

255

muffler (Type-A) has the lowest (2.76 by % vol) amount of CO₂ emissions from pure gasoline

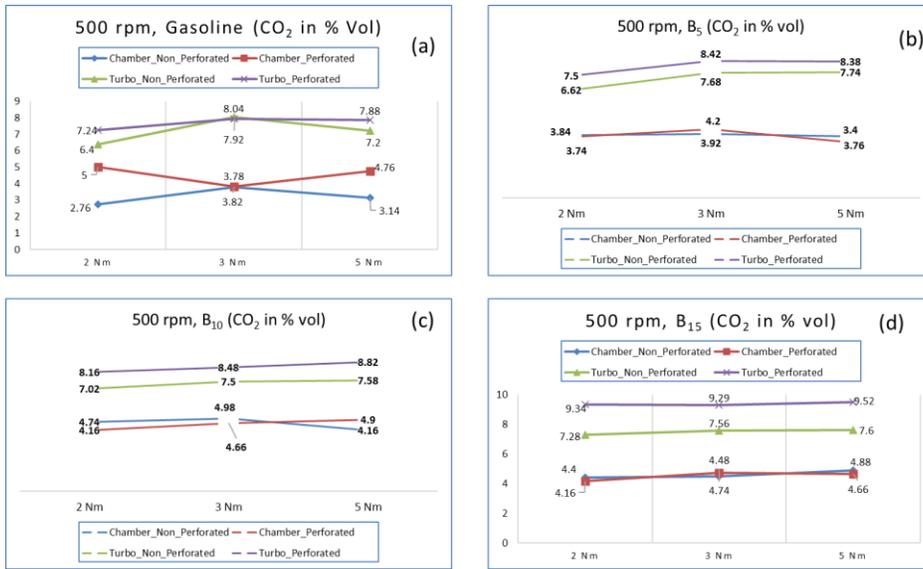
256

fuel at 2 Nm. The turbo perforated (Type-D) muffler has the highest CO₂ emissions (9.52 by %

257

vol). Perforation led to a 42% increase in CO₂ for all chambered type mufflers at 2 Nm and 5

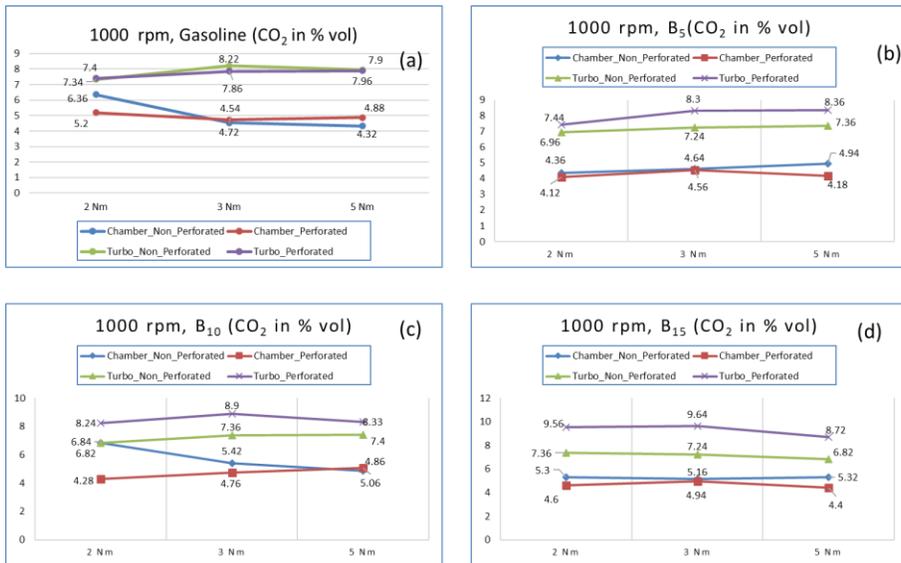
258 Nm, while using gasoline at 500 rpm. The effect of perforation in the chambered type muffler
 259 shows less difference in CO₂ emission at 500 rpm. For the turbo type muffler, running on pure
 260 gasoline fuel the difference is less, but for blended fuels (B₅, B₁₀ and B₁₅) larger differences of
 261 CO₂ emissions are observed, and more so in the case of the perforated turbo type (Type-D). The
 262 chambered type mufflers have almost half the level of CO₂ emissions as compared to turbo
 263 mufflers.



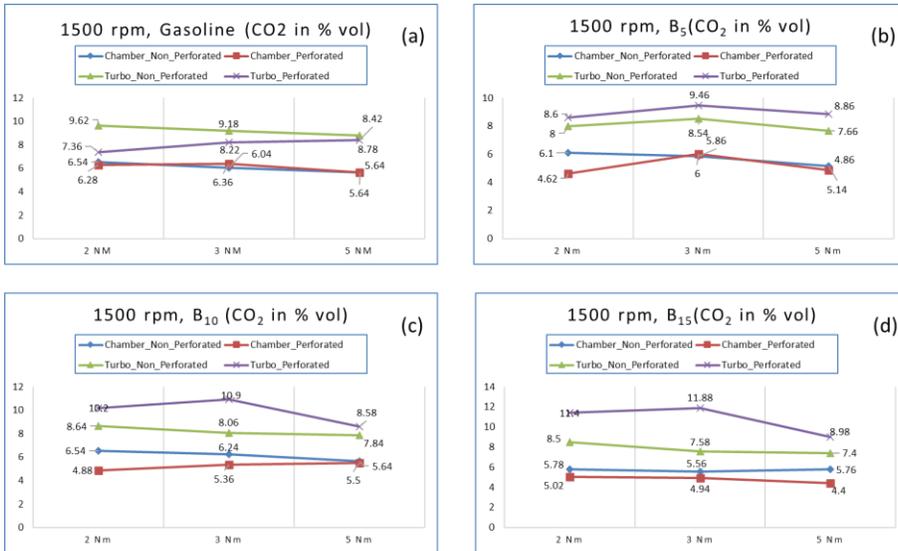
264 **Fig.6** CO₂ response to engine torque, (a) Gasoline at 500 rpm, (b) B₅ at 500 rpm, (c) B₁₀ at 500 rpm and
 265 (d) B₁₅ at 500 rpm

267 Figs. 7(a-d) show the CO₂ response to engine torque (2 Nm, 3 Nm and 5 Nm) at 1000 rpm. As
 268 the speed increases from 500 rpm to 1500 rpm, the CO₂ emissions increase in all cases. Lowest
 269 value of CO₂ emissions is 4.12 by % vol. at 2 Nm for the chambered type non-perforated muffler
 270 (Type-A). Similarly, at 1000 rpm the turbo perforated muffler (Type-D) shows maximum CO₂
 271 emissions of 9.64 by % vol at 3 Nm. The effect of perforation has again less effect in case of the

272 chamber type muffler at 3 Nm, which is 0.18,0.08, 0.66 and 0.22 for gasoline blends B₅, B₁₀ and
 273 B₁₅, respectively. Perforations lead to a maximum 59.3% increase in CO₂ emission in case of
 274 using the chamber type muffler with B₁₀ at 2 Nm. Furthermore, the turbo type muffler gives a
 275 maximum 29.9% increase in CO₂ emission B₁₅ at 2 Nm.



276
 277 **Fig.7** CO₂ response to engine torque, (a) Gasoline at 1000 rpm, (b) B₅ at 1000 rpm, (c) B₁₀ at 1000 rpm
 278 and (d) B₁₅ at 1000 rpm

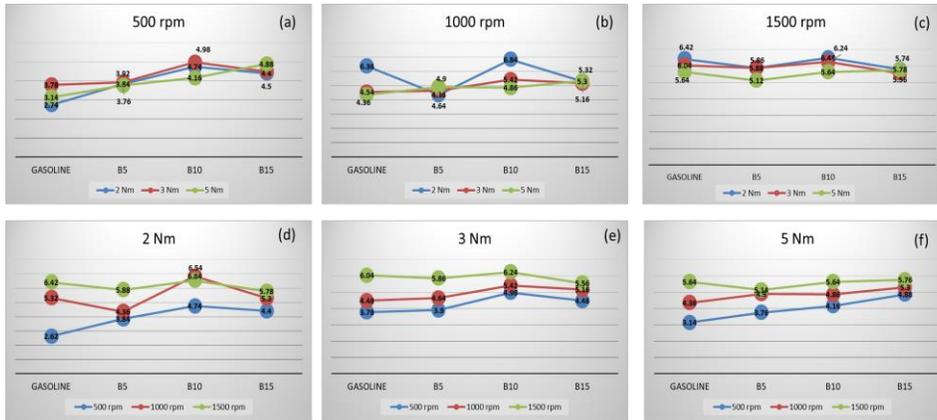


279

280 **Fig.8** CO₂ response to engine torque, (a) Gasoline at 1500 rpm, (b) B₅ at 1500 rpm
 281 and (d) B₁₅ at 1500 rpm

282 Fig. 8 (a-d) show CO₂ response to engine torque at 1500 rpm for Gasoline, B₅, B₁₀ and B₁₅
 283 respectively. The lowest CO₂ emissions at this speed are observed to be 4.28 by % vol. for the
 284 type-A muffler at 2 Nm. The highest CO₂ formation (11.88 by % vol) occurs at 3 Nm for the
 285 type-D muffler. For the turbo type muffler, the effect of perforation enhances the CO₂ emissions
 286 by 56.7 % at 3 Nm for B₁₅, while for the chamber type muffler, the effect of perforation
 287 enhances CO₂ emissions by 34.01% for B₁₀ at 2 Nm. At 3 Nm, for the Chamber type, muffler
 288 perforation has negligible effect on CO₂ emissions (0.32, 0.14, 0.88 and 0.62) by % vol. for all
 289 fuels (Gasoline, B₅, B₁₀ and B₁₅). The effect of perforations in the turbo type muffler has the
 290 highest impact on CO₂ emission with (0.96, 0.92, 2.84 and 4.3) by % vol. for (Gasoline, B₅, B₁₀
 291 and B₁₅)

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292

293 **Fig.9** CO₂ response to fuel type at: (a) at 500 rpm, (b) 100 rpm, (c) 1500 rpm, (d) 2 Nm, (e) 3 Nm and (f)
 294 5 Nm

295 Figs. 9(a-f) show the CO₂ response to fuel change (Gasoline, B₅, B₁₀ and B₁₅). When gasoline is
 296 replaced with B₅, the emission levels are found to increase (0.14,0.36, 0.24, -1.06, 0.16 and 0.54)
 297 by % vol. at (500 rpm, 1000 rpm, 1500 rpm, 2 Nm, 3 Nm and 5 Nm). Similarly, when gasoline is
 298 replaced by blend B₁₀, the emission levels are also found to increase (1.2,0.88, 0.6, 1.52, 0.94
 299 and 0.5) by % vol. at (500 rpm, 1000 rpm, 1500 rpm, 2 Nm, 3 Nm and 5 Nm). Furthermore,
 300 when gasoline is replaced by blend B₁₅, the emission levels are found to increase (-0.1, -0.1, -
 301 0.46, -1.36, -0.26 and 0.44) by % vol. at (500 rpm, 1000 rpm, 1500 rpm, 2 Nm, 3 Nm and 5 Nm).

302 *3.2 Muffler CFD Simulation results and discussions*

303 In order to understand the muffler exhaust performance, we have decided to carry out
 304 computational fluid dynamics simulation in Ansys fluid, Prior to such analysis, the solid model
 305 of all four mufflers are prepared, conforming to the geometries of fabricated ones. (Mishra et al.
 306 2018) explained the step-by-step build-up procedure for solid modelling using CATIA. Later,
 307 such models are imported to ANSYS fluid, which are compatible to it. The very first step in

308 ANSYS is to create mesh model of the mufflers. Details of the mesh such as number of
 309 elements, number of nodes, element size, element type etc. are automatically selected by ANSYS
 310 work bench. Here the element type auto-selected are tetrahedral with 180 curvature normal
 311 angle. The first step of CFD is pre-processing, which includes defining inlet and outlet surface of
 312 the control volume of muffler. Table 4 shows the parameter required under inlet boundary
 313 conditions, that includes densities, enthalpies and viscosities for gasoline, B₅, B₁₀ and B₁₅. The
 314 table 5 shows the velocity variation at inlet for all muffler design at gasoline, B₅, B₁₀ and B₁₅ use.
 315 Table 6 and 7 show the other input parameter and the mass fraction at inlet respectively as drawn
 316 from the emission measurement. The mesh model along with input parameters are loaded in the
 317 solver for output data generation.

318 **Table 4** Parameters under inlet boundary conditions.

<i>Models/Parameters</i>	<i>Inlet Boundary</i>			
	<i>Pure</i>	<i>B5</i>	<i>B10</i>	<i>B15</i>
Density (kg/m ³)	1.021	1.0271	1.0170	1.024
Enthalpy (j/kg)	163675	156911	160984	159136
Viscosity (kg/m-s)	0.0000172	0.0000172	0.0000172	0.0000172

319 **Table 5** Velocity (m/s) variation at the inlet.

<i>Models/Blends</i>	<i>Pure</i>	<i>B5</i>	<i>B10</i>	<i>B15</i>
Type-A&B	0.1051	0.1084	0.1144	0.119
Type-C&D	0.1050	0.1085	0.1144	0.119

320

321 **Table 6** Other input parameters

<i>Parameter</i>	<i>Pure</i>	<i>B5</i>	<i>B10</i>	<i>B15</i>
Temperature(°C)	363	357	360	359

Mass Flow Rate (Kg/sec)	0.00052	0.00055	0.00057	0.0006
HT Coefficient (W/m ² K)	35	42	46	52

322

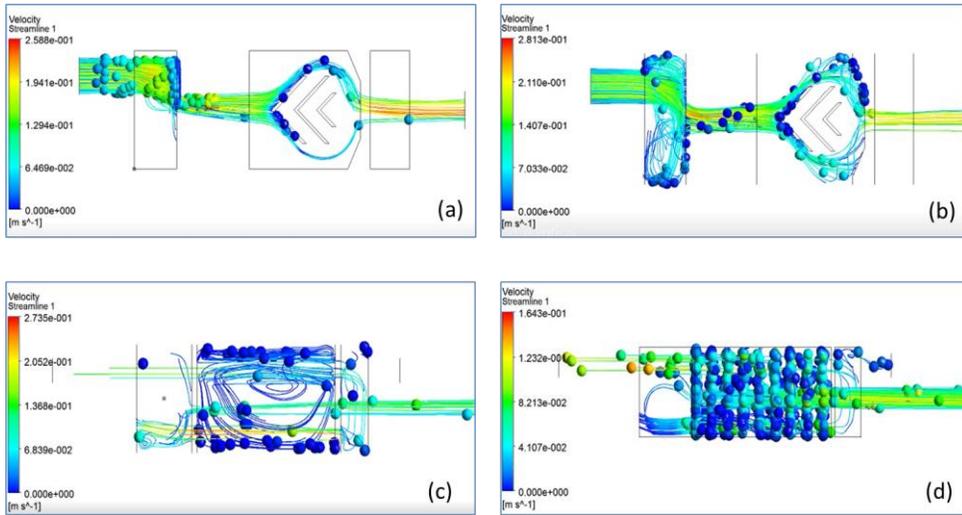
323 **Table 7** Mass fraction at inlet (As input parameter).

<i>Cases/Gases</i>	<i>CO</i>	<i>NO_x</i>	<i>HC</i>	<i>CO₂</i>	<i>O₂</i>
Pure Gasoline	0.027544	0.00011	0.000064	0.06241	0.17668
B ₅	0.0331	0.0001924	0.000082	0.062602	0.17531
B ₁₀	0.027301	0.00014208	0.0000610	0.06057	0.17975
B ₁₅	0.026771	0.00021227	0.0000556	0.077116	0.16402

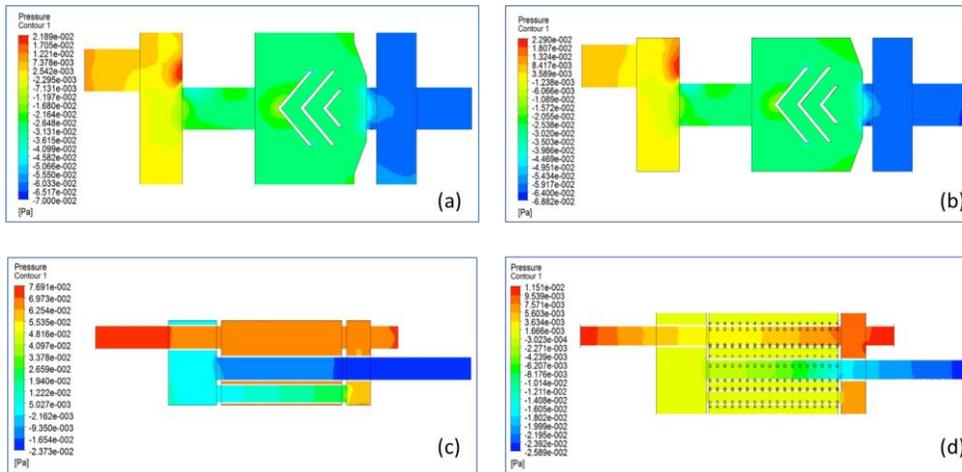
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325 The solver used in this analysis is a pressure-based solver, where heat transfer is address through
 326 energy model and assumed that flow of heat occurs from hot exhaust gas to the walls of the
 327 muffler. The mixture of exhaust gases considered at muffler inlet are NO_x, CO, CO₂ and HC.
 328 Mass fraction of each constituent exhaust gas was calculated in terms of SI unit. Further enclosed
 329 walls are considered stationery with non-slip specific boundary with zero diffusion ability. As
 330 the exhaust gas moves faster, geometrical obstruction opposes the flow and creates back
 331 pressure. The mesh convergence and grid independent tests were carried out to ensure correction
 332 grid formation and that of mesh size. The figures (10-13) shows muffler performance parameters
 333 for all four models. Fig. 10 shows velocity streamline from inlet to outlet, maximum velocity of
 334 0.28 m/s occurs at type-B muffler, while lowest 0.1643 m/s observed in case of type-D muffler.

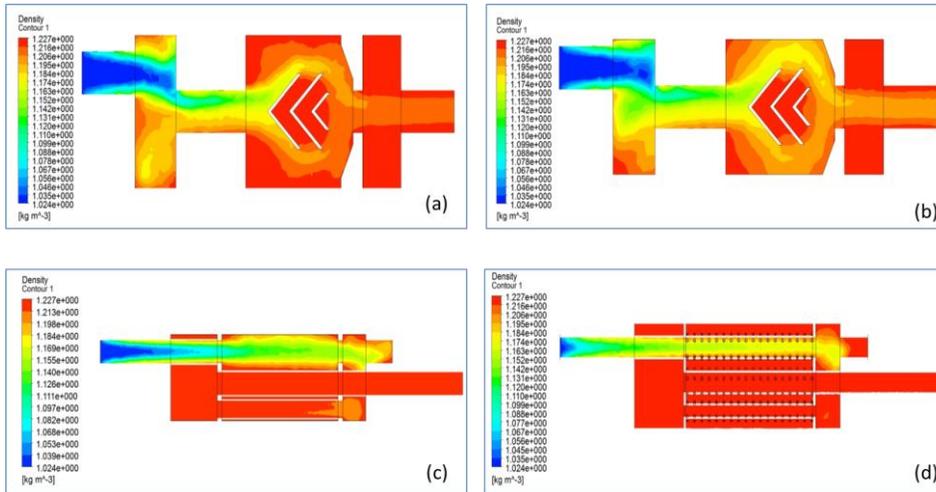
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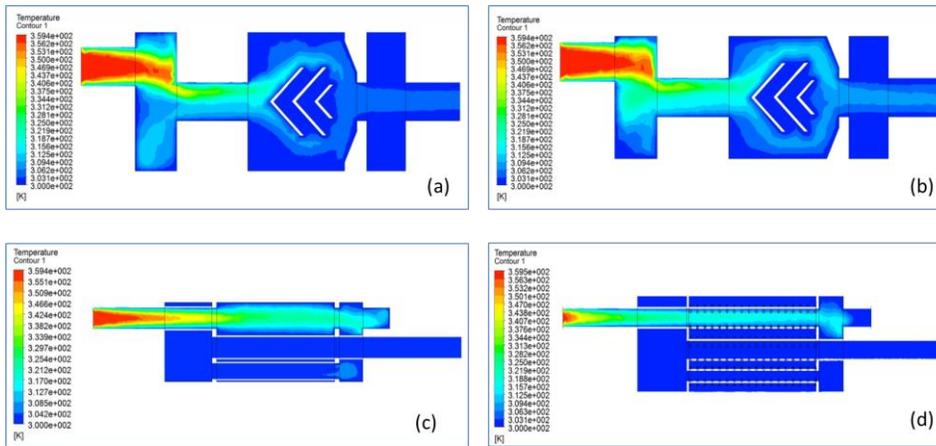
336
 337 **Fig.10** Velocity stream line from CFD simulation of: (a) Type-A muffler, (b) Type-B muffler, (c) Type-C
 338 muffler and (d) Type-D muffler



339
 340 **Fig.11** Back pressure from CFD simulation of: (a) Type-A muffler, (b) Type-B muffler, (c) Type-C
 341 muffler and (d) Type-D muffler



342
 343 **Fig.12** Emission gas density from CFD simulation of: (a) Type-A muffler, (b) Type-B muffler, (c) Type-
 344 C muffler and (d) Type-D muffler



345
 346 **Fig.13** Emission gas temperature from CFD simulation of: (a) Type-A muffler, (b) Type-B muffler, (c)
 347 Type-C muffler and (d) Type-D muffler

348 Fig. 11 shows the back-pressure mapping of the four different mufflers, result obtained from
 349 CFD simulation. The highest back pressure of 0.0769 Pa observed for turbo non-perforated

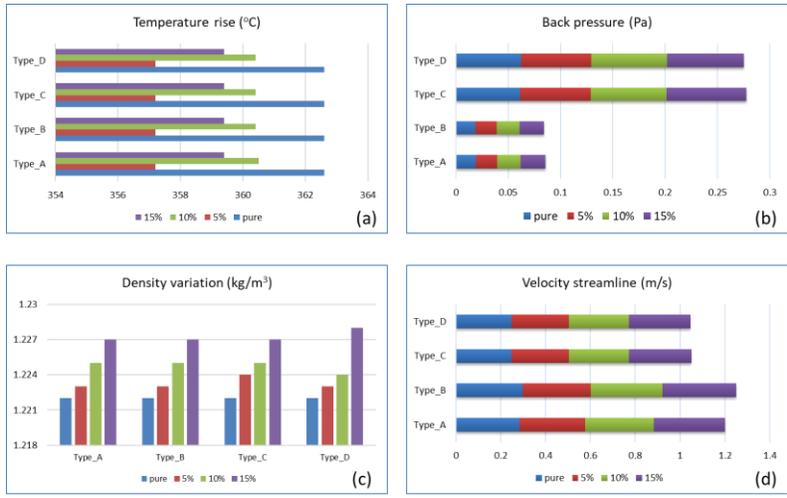
350 muffler, while the lowest of 0.0151 Pa found in case of turbo-perforated one. The fig 12 shows
 351 the density distribution in the muffler control volume. Not much variation in the density is
 352 observed. Fig. 13 shows the temperature distribution of the four different models. The
 353 distribution of temperature is more in the turbo mufflers compared to chambered type mufflers.
 354 In table 8 the key exhaust performance parameters are summarized.

355

356 **Table 8** Summary of key exhaust performance parameters.

Muffler detail and blend detail	Maximum value of velocity in (m/s)	Maximum value of density in (kg/m ³)	Maximum value of back pressure in (Pa)x10 ⁻²	Maximum value of exhaust temperature in (°K)
Type-A&B muffler using pure gasoline	0.24	1.22	6.2	362
Type-C&D muffler using pure gasoline	0.146	1.25	1.4	362
Type-A&B muffler using B5	0.25	1.22	6.7	357
Type-C&D muffler using B5	0.152	1.27	1.49	357
Type-A&B muffler using B10	0.25	1.22	7.3	360
Type-C&D muffler using B10	0.158	1.26	1.5	360
Type-A&B muffler using B15	0.27	1.22	7.7	359
Type-C&D muffler using B15	0.164	1.30	1.65	359

357



358

359 **Fig.14** Bar chart of muffler performance, (a) Temperature rise, (b) back pressure variations, (c) density
 360 variation and (d) velocity stream line

361 Fig. 14(a-d) show the performance of the mufflers, which include surface temperature, back
 362 pressure, density and stream line velocity respectively. Surface temperature (fig.14-a) is almost
 363 independent to the muffler design. This is highest in case of pure gasoline fuel and lowest in the
 364 case of blend B₁₀. The back pressure (fig. 14-b) is the higher in case of the chambered type
 365 (Type-A and Type-B) mufflers as compared to the turbo types (Type-C and Type-D). Fig. 9-c
 366 shows the density variation of exhaust gas, which increases with % increase of methanol in the
 367 blend. As we proceed to the higher order of the blend, the combustion improves and yields more
 368 dense emission constituents. As shown in fig. 14-d, the velocity streamline is higher, in case of
 369 the chamber type design, as the path is simple, while in case of the turbo design the path is
 370 circulatory and more complex.

371 **5. Sustainability analysis**

Commented [R18]: Don't we need to ask why?

372 The sustainability of the design modification and fuel change is worth discussing here. As
373 mentioned earlier, the maximum blending possible is up to 15%. Higher order of methanol
374 content is not encouraged in this analysis, and the reason is the toxicity of methanol content and
375 also the flameless combustion. Secondly, methanol is easily available and it can be produced in
376 the Petro-chemical facilities with minimal additional investment. There is absolutely no change
377 in engine infrastructure required for such small fuel changes. The direct advantage is that the
378 CO₂ formation is reduced as compared to pure gasoline. This direct benefit is available for no
379 changes in engine structure or materials and minimal costs in fuel preparation as no separate
380 blending facility is needed. Therefore, the improved impact on the environment from the
381 reduction of engine emissions of CO₂ outweighs the potential cost of blending and providing this
382 fuel directly to the pump for vehicle users. Furthermore, it can potentially reduce the demand on
383 pure gasoline by 15% volume. Which could result in financial benefits for the oil companies. It
384 has the potential of creating a new ‘Methanol Economy’ which can create opportunities for
385 economic prosperity.

386 **6. Conclusion**

387 This study has investigated how it is possible to improve the CO₂ emissions from a gasoline
388 engine by using blended gasoline-methanol fuels without negatively impacting the engine
389 performance, or having a hugely detrimental effect on the engine structure.

390 In addition to the gasoline-methanol blends, the design of the muffler have also been investigated
391 for their effect on CO₂ emissions. The key findings are summarized as:

- 392 • Chambered type non-perforated muffler (Type-A) is best among all designs for reduced
393 CO₂ emissions.
- 394 • Turbo perforated (Type-D) muffler has most CO₂ emissions compared to all designs.

Commented [RI9]: We need to qualify this last statement as there has been no investigation into the materials or components of the engine.

- 395 • For chambered type muffler, effect of perforation is negligible at 3 Nm for all range of
396 engine speed (500, 1000,1500) rpm.
- 397 • Effect of perforation is maximum at 3 Nm for turbo type muffler.
- 398 • At constant speed, the CO₂ emissions are higher for lower torque and at constant torque,
399 CO₂ emissions are is lower for lower speed.

400 Such minor modifications has immediate implication to the automotive sectors and fuel
401 manufacturers. Introduction of methanol could reduce burden on petroleum reserves with the
402 benefit of reduced CO₂ emissions. The limitation of the current analysis is that the methanol is
403 restricted to 15% in the blend. This is to eliminate the toxic effect and rapid degradation of
404 engine components.

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