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Journal article

### **Fungal solubilisation and subsequent microbial methanation of coal processing wastes**

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1 **Fungal solubilisation and subsequent microbial methanation of coal**  
2 **processing wastes**

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## 18 **Abstract**

19 Large quantities of rejects from coal processing plants are currently disposed of as waste piles  
20 or in ponds and rivers, resulting in environmental concerns including pollution of rivers, and  
21 ground and surface water contamination. This work investigates for the first time, a two-stage  
22 microbial process for converting coal processing wastes (coal rejects) to methane, involving  
23 (1) fungal solubilisation of coal rejects and (2) microbial methanation of the solubilised  
24 products. *Phanerochaete chrysosporium*, *Trichoderma viride* and *Neurospora discreta* were  
25 screened for their ability to solubilise coal rejects. *N. discreta* was found to be the most  
26 suitable candidate based on the extent of bio-solubilisation, laccase activity, and reversed-  
27 phase high-performance liquid chromatography (RP-HPLC) analysis. Bio-methanation of  
28 fungal-solubilised coal rejects was carried out in mesophilic anaerobic reactors with no  
29 additional carbon source, using inoculum from an anaerobic food digester. Coal rejects  
30 solubilised by *N. discreta* produced 3 to 6-fold higher methane compared to rejects solubilised  
31 by the other two fungi. No methane was produced from untreated coal rejects,  
32 demonstrating the importance of the fungal solubilisation stage. A total of 3.7 mmol of  
33 methane was generated per gram of carbon in 15 days from *N. discreta*-solubilised coal  
34 rejects. This process offers a timely, environment-friendly, and sustainable solution for the  
35 treatment of coal rejects and the generation of value-added products such as methane and  
36 volatile fatty acids.

37

## 38 **Keywords**

39 Coal processing waste; coal rejects; coal fungal solubilisation; coal bio-methanation;

40 *Neurospora discreta*

## 41 **1. Introduction**

42 Coal remains one of the most significant energy resources around the world with global  
43 consumption of nearly 8000 Mt per year [1]. Continuing to meet this demand despite steadily  
44 depleting deposits of high-rank coal has led to the mining of low-value coals such as sub-  
45 bituminous coal, lignite, and high-ash bituminous coal, which are abundant in North America,  
46 Europe and Asia-Pacific regions [2]. In recent years, there has been a steady increase in the  
47 mining and utilisation of low-rank and low quality coals in some of the largest coal-producing  
48 countries such as China, India and the USA.

49 Low-rank coals have high ash and moisture content and low thermal efficiencies compared to  
50 high-rank coals such as anthracite and therefore need to be subjected to coal beneficiation  
51 or upgradation to reduce ash content before being used for power generation [3–5].  
52 However, as the process of separating ash from coal is particularly challenging for low-rank  
53 coals, nearly 30-40% of coal is rejected in coal processing plants, resulting in millions of tonnes  
54 of coal processing waste (coal rejects) every year [4,6–8]. Depending on the beneficiation  
55 process, dry coal rejects are typically disposed of as solid waste piles while coal reject slurries  
56 are discarded in rivers (especially in India) or within embankments or ponds [9]. These  
57 disposal methods have led to serious environmental issues including pollution of rivers,  
58 ground and surface water contamination from reject area leachate, and fugitive emission of  
59 dust [6,8,10].

60 Coal rejects typically contain more than 50-60% ash but also contain up to 15% carbon and  
61 other combustibles that can potentially be utilised [11]. Recent research has explored the  
62 utilisation of coal rejects in fluidised bed combustion [10] and the recovery of clean coal from  
63 washery rejects using physical and chemical methods [6,11]. However, high inputs of energy,

64 the need for high-strength chemicals and low recoveries from these processes currently  
65 render these methods largely non-viable.

66 Can a biological process for treating coal rejects offer a sustainable and environment-friendly  
67 solution to these challenges?

68 Although studies on biodegradation of coal processing wastes are limited, filamentous fungi  
69 such as *Trichoderma viride* and *Phanerochaete chrysosporium* and certain aerobic bacteria  
70 have been shown to degrade low-rank coals such as lignite [12–18]. These microorganisms  
71 contain multiple ligninolytic and other oxidative and reductive enzymes that carry out the  
72 depolymerisation and bio-solubilisation of the coal structure, which is similar to that of lignin  
73 for low-rank coals [13]. Studies with lignite have shown the degradation of the coal matrix to  
74 lower molecular weight aromatic and aliphatic compounds that could potentially be  
75 converted to value-added products [17,19,20].

76 A different set of studies has explored the microbial generation of methane from coal, arising  
77 from the recent understanding of the role of microorganisms in coalbed methane generation  
78 – originally considered to be a purely thermogenic process [21]. Microbial methane  
79 production from sub-bituminous coal and lignite has been demonstrated at lab-scale,  
80 although this is a relatively slow process taking more than 60-70 days and even up a few  
81 hundred days in some cases[22–24].

82 These bio-solubilisation and bio-methanation studies independently demonstrate that low-  
83 rank coal can be microbially converted to either liquid products or methane, although the  
84 significantly long process durations remain a challenge in the case of methane production. A  
85 gap exists in evaluating a combined approach of bio-solubilisation and bio-methanation, to  
86 improve the digestibility of the coal matrix for methane production. Furthermore, till date no

87 similar studies have been reported on coal rejects. It is useful to note the differences between  
88 coal rejects and low-rank coal as potential substrates for microorganisms. Coal rejects have  
89 significantly higher ash content and lower carbon content compared to low-rank coal. Lignite  
90 for instance, contains about 60-70% carbon [21] while coal rejects contain less than 20%  
91 carbon. The structure of coal rejects is also likely to be less recalcitrant than that of coal,  
92 making it easier to degrade. This, coupled with the fact that coal rejects are currently a wasted  
93 resource, makes coal rejects a promising substrate for microbial methane production.

94 The present work is based on the hypothesis that coal rejects can be converted to methane  
95 using a two-stage biological process: (1) fungal solubilisation of coal rejects to produce  
96 simpler, water-soluble degradation products and (2) bio-methanation of the solubilised  
97 products using anaerobic microorganisms. Considering the environmental hazards posed by  
98 inappropriate disposal of these rejects, and the large quantities in which they are produced,  
99 this process offers a timely, sustainable, and environment-friendly solution for the treatment  
100 of coal rejects, as well as the extraction of a valuable fuel in the form of methane.

101

## 102 **2. Materials and Methods**

### 103 **2.1 Coal Rejects**

104 Samples of coal-washery rejects were kindly supplied by Ardee Hi-Tech Pvt Ltd,  
105 Visakhapatnam, India. The particle size and minimum ash content of the coal rejects were 0.2  
106 mm and 75% respectively. The coal rejects were sourced from Talcher coal mines, India, which  
107 contain sub-bituminous coal with high ash content.

### 108 **2.2 Fungal solubilisation of coal rejects**

109 Three fungal species were screened for their ability to solubilise the coal rejects.  
110 *Phanerochaete chrysosporium* (NCIM 1197) and *Trichoderma viride* (NCIM 1060) were  
111 obtained from National Collection of Industrial Microorganisms, Pune, India. These two fungi  
112 were selected for their reported ability to degrade low-rank coal [13,15,18]. The third fungus,  
113 *Neurospora discreta* was previously isolated from a Subabul wood tree and was selected for  
114 its ability to produce ligninolytic enzymes and degrade lignin [25,26]. All fungi were sub-  
115 cultured on potato dextrose agar (PDA) plates and at 2-8°C until further use.

116 Fungal solubilisation of coal rejects was carried out as submerged fermentation in 250 mL  
117 Erlenmeyer flasks containing 100 mL Vogel's minimal medium [27] with 1 g sucrose and 1 g  
118 coal rejects. After sterilisation and cooling, 0.1% biotin solution was added to each flask, and  
119 the flasks were inoculated in triplicate with a spore suspension of each fungal species. To  
120 prepare the spore suspension, cells were scraped from the agar plates and filtered through a  
121 muslin cloth and the spore suspension obtained was added to each flask to get a final  
122 concentration of 0.2 million spores per mL. All flasks were then incubated in a shaker  
123 incubator at 30°C and 100 rpm for 14 days. Un-inoculated coal rejects in Vogel's medium were  
124 set up as controls.

### 125 **2.3 Analysis of solubilised products, enzyme activity, protein content and dry weight**

126 Liquid samples were taken from each flask at regular intervals, centrifuged to remove solids,  
127 and analysed using RP-HPLC on a C-18 column, using a mixture of acetic acid and acetonitrile  
128 as the mobile phase using the method described elsewhere [25]. Alkali lignin (low sulphonate  
129 Kraft lignin, Sigma Aldrich) was used as a reference standard. Controls (coal rejects without  
130 fungal treatment) and media blanks were also run using the same method.

131 Liquid supernatant obtained after centrifugation of samples from each flask was analysed for  
132 laccase activity based on oxidation kinetics of ABTS. Absorbance of the blue-green radical  
133 formed by the enzymatic oxidation of ABTS was measured at 420 nm and enzyme activity was  
134 calculated as the amount of enzyme forming  $1 \mu\text{M}\cdot\text{min}^{-1}$  of product, using an extinction  
135 coefficient ( $\epsilon_{420}$ ) of  $36000 \text{ L}\cdot\text{mol}\cdot\text{cm}^{-1}$  [26].

136 Protein content in the solid fraction was used as an indirect measure of cell growth. For this,  
137 a known mass of the solid fraction was subjected to protein extraction by incubating with  
138 Radio-Immunoprecipitation Assay (RIPA) lysis buffer containing 1mM phenyl methyl  
139 sulphonyl fluoride (PMSF) (both from Sigma Aldrich) for one hour at room temperature with  
140 manual glass bead vortexing every 15 minutes. 10 ml of buffer was used per gram of solid.  
141 The lysate was then centrifuged at  $2500 \times g$  for 5 minutes and the protein content in the  
142 supernatant was estimated using the Folin-Lowry method [28].

143 Dry weight of the residual coal was obtained after drying the solid fraction at  $103.5^\circ\text{C}$  in an  
144 oven until constant weight was achieved.

## 145 **2.4 Bio-methanation of fungal-solubilised coal rejects**

### 146 *2.4.1 Batch reactor set-up*

147 A schematic representing the fungal solubilisation and bio-methanation experiments is shown  
148 in figure 1. Batch bio-methanation studies were carried out in 250 mL serum bottles using the  
149 fungal-solubilised coal samples. In the figure and description below, the letters N, P, T denote  
150 coal rejects subjected to bio-solubilisation by *N. discreta*, *P. chrysosporium*, *T. viride*  
151 respectively and C denotes the control (coal rejects without fungal treatment). Each reactor  
152 contained 40% by volume of the bio-solubilised coal and 45% modified Barker's medium



153 [29,30]. The medium contained 20 g.L<sup>-1</sup> CaCO<sub>3</sub>, 1.0 g.L<sup>-1</sup> NH<sub>4</sub>Cl, 0.1 g.L<sup>-1</sup> MgCl<sub>2</sub>.6H<sub>2</sub>O and 0.4  
154 g.L<sup>-1</sup> K<sub>2</sub>HPO<sub>4</sub>.3H<sub>2</sub>O but with no additional carbon source. All reactors were purged with  
155 nitrogen for 5-7 minutes with a long needle while simultaneously boiling the medium to  
156 remove oxygen and then sealed with rubber septa and aluminum crimp seals to maintain an  
157 anaerobic environment. After autoclaving and cooling, sterile 0.5mM Na<sub>2</sub>S was added. Each  
158 reactor was then inoculated with 15% inoculum from a mesophilic anaerobic digester for food  
159 waste, kindly supplied by BITS Pilani, Goa campus. Water was added to the control in place of  
160 the inoculum. All reactors were incubated at 37°C. Methane concentration in the headspace  
161 and volatile fatty acids (VFA) in the liquid samples were analysed as described below.

#### 162 *2.4.2 Determination of volumetric methane production*

163 To determine the volume of methane produced in coal rejects solubilised by *N. discreta* (N-1,  
164 Fig. 1), the reactor was sealed using a rubber stopper with a tube to allow the headspace gas  
165 to exit (instead of the crimp). The gas passed through a solution of 0.1 M calcium hydroxide  
166 solution to strip CO<sub>2</sub> and into an inverted measuring cylinder filled with water in a water  
167 trough. The volume of methane-enriched gas was determined by the volume of water  
168 displaced in the measuring cylinder.

#### 169 *2.4.3 Effect of media addition*

170 In a separate study (N-2, Fig. 1), once the methane gas production slowed down in the batch  
171 reactors, 45% degassed Barker's medium was added to 55% of the broth from the batch  
172 reactor (N) under anaerobic conditions. As before, no additional carbon source was added.  
173 Liquid samples were withdrawn anaerobically for VFA analysis, and the headspace gas was  
174 analysed for methane as described below.

#### 175 2.4.4 Determination of methane gas concentration and VFA

176 Methane gas in the headspace was measured using a portable biogas analyser (BIOGAS 5000,  
177 Geotech, India), connected to a needle to pierce the rubber septa.

178 Liquid samples from the anaerobic reactors were centrifuged at 10,000 x g for 10 minutes and  
179 the supernatant was put through a 3-point titration for pH 5.0, 4.3, and 4.0. Total VFA was  
180 calculated according to the following formula [31,32]:

$$181 \text{ Total VFA (mg. L}^{-1}\text{)} = \left[ 131,340 * (V_{pH4.0} - V_{pH5.0}) * \frac{N_{H_2SO_4}}{V_s} \right] - \left[ 3.08 * V_{pH4.3} * \frac{N_{H_2SO_4}}{V_s} * 1000 \right] - 10.9$$

182

183 In the above formula,  $V_{pH4.0}$ ,  $V_{pH4.3}$ , and  $V_{pH5.0}$  are the volumes (in mL) of acid added until pH  
184 of 4.0, 4.3, and 5.0 are achieved, respectively.  $V_s$  is the volume of the titration sample in mL  
185 and  $N_{H_2SO_4}$  is the normality of sulphuric acid.

186

### 187 3. Results and Discussion

#### 188 3.1 Screening of fungal species for bio-solubilisation

##### 189 3.1.1 Extent of bio-solubilisation and laccase activity

190 The protein content in the solid biomass was similar for all three fungal species, indicating  
191 similar cell growth (Fig. 2). However, the mass of residual coal rejects varied based on the  
192 fungus indicating a difference in the extent to which the solid coal was solubilised in each  
193 case. At the end of 14 days, *N. discreta* resulted in a 55% reduction in the mass of coal rejects,  
194 which was the highest amongst the three species. *T. viride* resulted in the least reduction of  
195 approximately 25%.

196 This trend is further confirmed by the activity of laccase, which was the highest in the case of  
197 *N. discreta* followed by *P. chrysosporium* which showed significantly lower activity (Fig. 2).  
198 Laccases are one of the primary groups of enzymes responsible for de-polymerisation and  
199 bio-solubilisation of coal, owing to their low specificity and ability to break down both  
200 phenolic and non-phenolic structures [19,33]. Extracellular laccases have been reported in all  
201 three fungi tested [26,34,35], however, some studies have indicated intracellular, membrane-  
202 associated laccases in *T. viride* [36]. This could be one of the factors contributing to the  
203 absence of laccase activity in the *T. viride* samples. It is also likely that other ligninolytic  
204 enzymes were responsible for fungal solubilisation. However, the positive correlation  
205 between laccase activity and extent of fungal solubilisation in each case indicates the laccase  
206 played a significant role in the solubilisation of coal rejects.

### 207 3.1.2 Analysis of bio-solubilisation products

208 Bio-solubilisation of coal has been shown to occur via the breakdown of the hydrophobic coal  
209 matrix into simpler, water-soluble (“liquified”) products [37,38]. In the present study, fungal  
210 bio-solubilisation of coal rejects resulted in the production of polar degradation products as  
211 confirmed by RP-HPLC chromatograms of the liquid samples (Fig. 3). Owing to the structural  
212 similarities between lignite and lignin [13], it can be expected that solubilisation of coal would  
213 result in products similar to soluble lignin. Therefore, water-soluble alkali lignin was used as  
214 the reference standard.

215 Each fungal species used for bio-solubilisation produced a different profile of degradation  
216 products. As bio-solubilisation progressed from day 7 to 14, coal rejects treated with *N.*  
217 *discreta* and *P. chrysosporium* showed a decrease in product heterogeneity (number of peaks)  
218 and a slight increase in polarity (based on retention time) (Fig 3a, b, d, e). Treatment with *T.*

219 *viride* resulted in no significant peaks on day 7 (Fig. 3c), indicating a slower degradation  
220 compared to the other two cases.

221 On day 14, coal rejects treated with *N. discreta* produced a single larger peak at a retention  
222 time (RT) close to 2.6 minutes (Fig. 3d), indicating the presence of a highly polar product  
223 similar to the soluble lignin standard (Fig. 3g). Solubilisation by *P. chrysosporium* and *T. viride*  
224 resulted in multiple smaller peaks (Fig. 3e, f). The coal control (without fungal treatment)  
225 sample consistently had a few small peaks, all below an intensity of 5 mAU.

226 A comparison of the areas under the curve (AUC) corroborates the observation from dry  
227 weights and enzyme activities that *N. discreta* resulted in the highest extent of bio-  
228 solubilisation, and *T. viride* the lowest (Fig. 4). In all cases the total AUC increased from day 7  
229 to day 14 indicating the progress of bio-solubilisation with time.

### 230 **3.2 Production of methane and VFA**

231 In the batch bio-methanation studies, methane production from coal rejects treated with *N.*  
232 *discreta* (reactor N, Fig.1) increased steadily till day 15, after which the rate of increase slowed  
233 down (Fig. 5). By day 23, the reactor headspace contained 60% methane which was six-fold  
234 higher than in reactor T and three-fold higher than in reactor P. Coal rejects without fungal  
235 treatment did not produce any methane in the period tested. This can be compared to studies  
236 reported with low-rank coal wherein methane production did not commence until after  
237 approximately 60 days [22–24].

238 Figure 5 in conjunction with figure 4, highlights the importance of the first stage in methane  
239 production and shows a positive effect of the extent of fungal solubilisation of coal rejects on  
240 methane production. This can be explained by the fact that the products of bio-solubilisation

241 are simpler structures that are easier to utilise by methanogens. Moreover, the polar nature  
242 of these products (as seen from the RP-HPLC chromatograms) significantly improves  
243 accessibility to the microorganisms compared to the highly hydrophobic coal particles.

244 VFA at harvest showed the opposite trend to methane production with 3-fold higher VFA  
245 production seen in coal rejects treated with *T. viride* compared to *N. discreta* as seen (Fig 4A).

246 VFAs are intermediate products in the methanogenic pathway, arising from the hydrolysis of  
247 the substrate and serving as precursors to methane formation. Therefore, a high  
248 concentration of methane, as in the case of *N. discreta*, and a relatively low residual VFA  
249 content in the reactor indicates the conversion of VFA to methane. Solubilisation by *P.*  
250 *chrysosporium* resulted in lower methane but higher VFA compared to *N. discreta*.

251 Interestingly, the high VFA concentration in *T. viride*- treated samples indicates that the  
252 anaerobic consortium was able to metabolise the degraded and solubilised coal products to  
253 some extent, although this did not translate to methane production in the given time scale.  
254 Longer periods of solubilisation and bio-methanation could increase methane production in  
255 these cases.

256 As discussed previously, the methane production in *N. discreta* slowed down between days  
257 15 and 23, increasing by only 2%. However, addition of fresh Barker's medium to the *N.*  
258 *discreta*-treated sample in the second stage (reactor N-2, Fig. 1) resumed methane  
259 production, which built up to over 35% in 10 days. This indicates that the slowdown in  
260 methane production in the first stage was not due to depletion of the carbon source (coal  
261 rejects) but due to depletion of other nutrients or a build-up of inhibitory by-products. It is to  
262 be noted that there was no residual methane on day 0 in the headspace as the substrate,  
263 culture and fresh medium were transferred to a new reactor. However, residual VFA from the

264 previous culture can still be seen in N-2 on day 0 and correlated well with the extent of dilution  
265 with fresh medium. In N-2, VFA dropped steadily with time reaching a value below 5 mg/L on  
266 day 10 once again confirming the conversion of VFA to methane.

267 From reactor N-1 (Fig. 1), 0.82 mmol (20 mL) of methane-enriched gas (>90% methane after  
268 CO<sub>2</sub> stripping) was produced per gram of coal rejects in 15 days. This amounts to  
269 approximately 0.74 mmol of methane per gram of coal rejects. Direct biogenic methane  
270 production from low-rank coal has been reported at much lower levels starting at 14-16 μmol  
271 per g of coal in 70 days, to approximately 0.2 mmol per gram in 63 days [23]. Wang et al [39]  
272 found that pre-treating lignite with pre-acclimatised aerobic sludge bacteria for 28 days  
273 followed by anaerobic digestion resulted in nearly 0.2 mmol of methane per gram of coal  
274 which was thrice the amount produced without pre-treatment. Considering the differences  
275 in carbon content between lignite and coal rejects, a better comparison would be in terms of  
276 methane per gram of carbon. At an average value of 65% total carbon in lignite [40,41], the  
277 highest methane production reported so far is 0.3 mmol per gram of carbon [23,39] which is  
278 significantly lower than the 3.7 mmol of methane per gram of carbon observed in the present  
279 study.

280

#### 281 **4. Conclusion**

282 This work demonstrates for the first time, a two-stage process for conversion of coal rejects  
283 to methane, involving fungal solubilisation followed by microbial methanation. Fungal  
284 solubilisation of coal rejects resulted in highly polar degradation products as analysed by RP-  
285 HPLC. Of the fungal species tested, *N. discreta* was found to be the most suitable candidate  
286 as it resulted in the highest extent of bio-solubilisation and consequently the highest amount

287 of methane production. Up to 60% methane was produced from coal rejects treated with *N.*  
288 *discreta* with a total of 3.7 mmol methane per gram of carbon in 15 days. This is more than  
289 ten-fold higher than the methane production reported from low-rank coals such as lignite.  
290 This two-stage process offers an environment-friendly solution for the conversion of coal  
291 rejects to methane. This process can also be extended to the upgradation of low-rank coals  
292 to avoid the use of high temperatures and pressures and generation of harmful by-products  
293 and gases. Optimisation of process conditions at the bio-methanation stage can lead to  
294 further improvement in methane yields. An analysis of individual VFAs produced can help  
295 identify other value-added products from coal rejects.

296

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299 Goa Campus for the inoculum for bio-methanation studies and BITS Pilani, Hyderabad  
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301

## 302 **6. Declarations**

### 303 **6.1 Funding**

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306 Vishakhapatnam, India.

### 307 **6.2 Conflicts of Interest**

308 The authors have no conflicts of interest to declare.

### 309 **6.3 Availability of data**

310 Data used during the present study can be requested from the corresponding author.

### 311 **6.4 Author contributions**

312 AA conceived and designed the experiments and wrote the manuscript. AS executed the  
313 experiments and collected data.

### 314 **6.5 Ethics approval**

315 Not applicable

### 316 **6.6 Consent to participate**

317 Not applicable

### 318 **6.7 Consent for publication**

319 Not applicable

320

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## 426 **Figure Captions**

### 427 **Fig. 1**

428 Schematic of fungal solubilisation and bio-methanation studies. N', P', T' represent fungal  
429 solubilisation by *N. discreta*, *P. chrysosporium*, *T. viride* respectively and C' represents the  
430 control. N, P, T, C represent bio-methanation of the coal rejects treated with *N. discreta*, *P.*  
431 *chrysosporium* and *T. viride* respectively and C represents untreated coal rejects. N-1 was set  
432 up to measure the volumetric methane production and N-2 was sub-cultured from N by  
433 adding fresh Barker's medium

### 434 **Fig. 2**

435 Mass of coal rejects before and after bio-solubilisation and protein content are depicted by  
436 bars and laccase activity is represented by the filled circles

437 **Fig. 3**

438 RP-HPLC chromatograms of liquid samples post fungal treatment of coal rejects. (a) *N.*  
439 *discreta* day 7 (b) *P. chrysosporium* day 7 (c) *T. viride* day 7 (d) *N. discreta* day 14 (e) *P.*  
440 *chrysosporium* day 14 (f) *T. viride* day 14 (g) Alkali lignin standard (h) Coal control (i) Media  
441 blank

442 **Fig. 4**

443 Total area under the curve (AUC) calculated from RP-HPLC chromatograms of liquid samples  
444 after treatment with *N. discreta*, *P. chrysosporium* and *T. viride*. The control contains un-  
445 inoculated coal rejects in media

446 **Fig. 5**

447 **a** Methane and VFA production from coal rejects treated with different fungi as a function of  
448 time

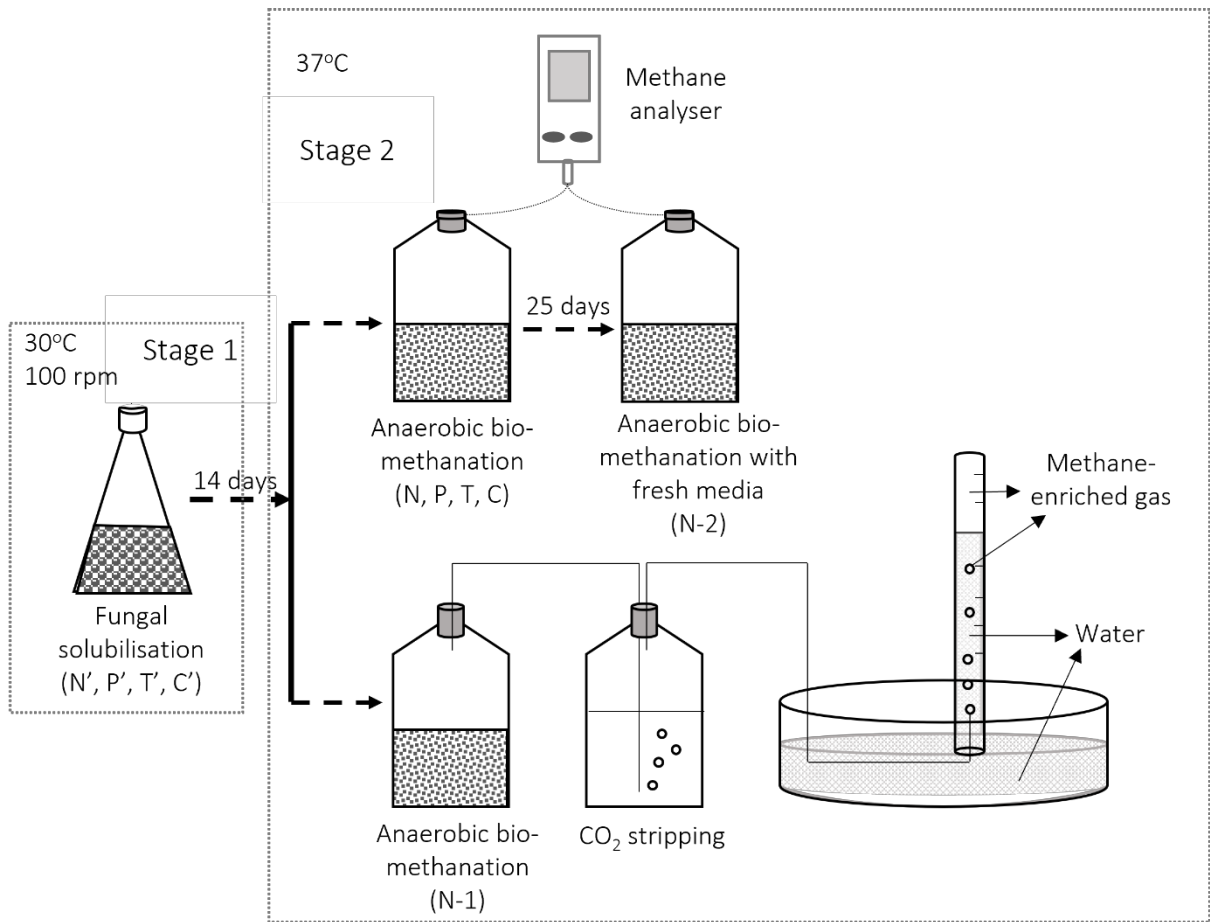
449 **b** Methane and VFA production after addition of fresh Barker's medium (reactor N-2)

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Figures

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Fig. 1

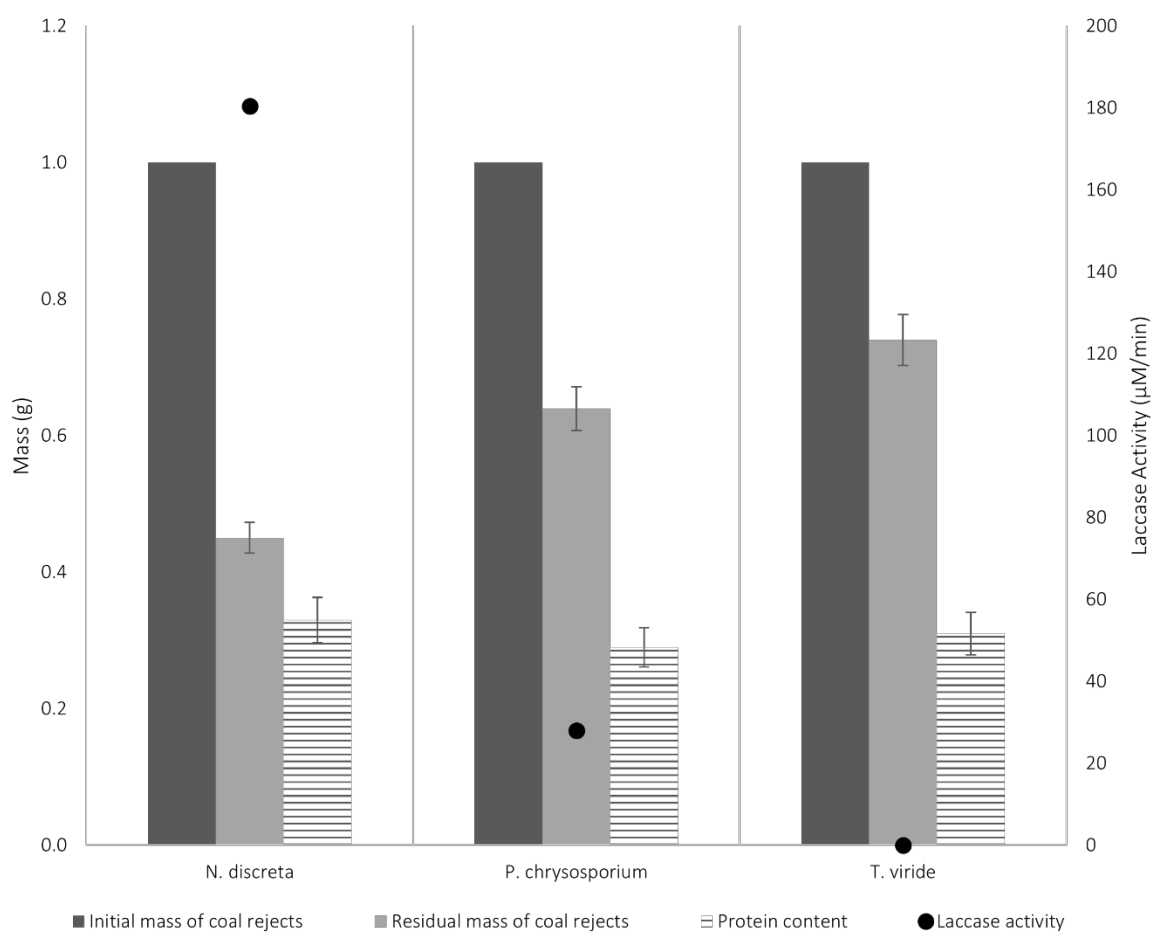


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466 **Fig. 2**

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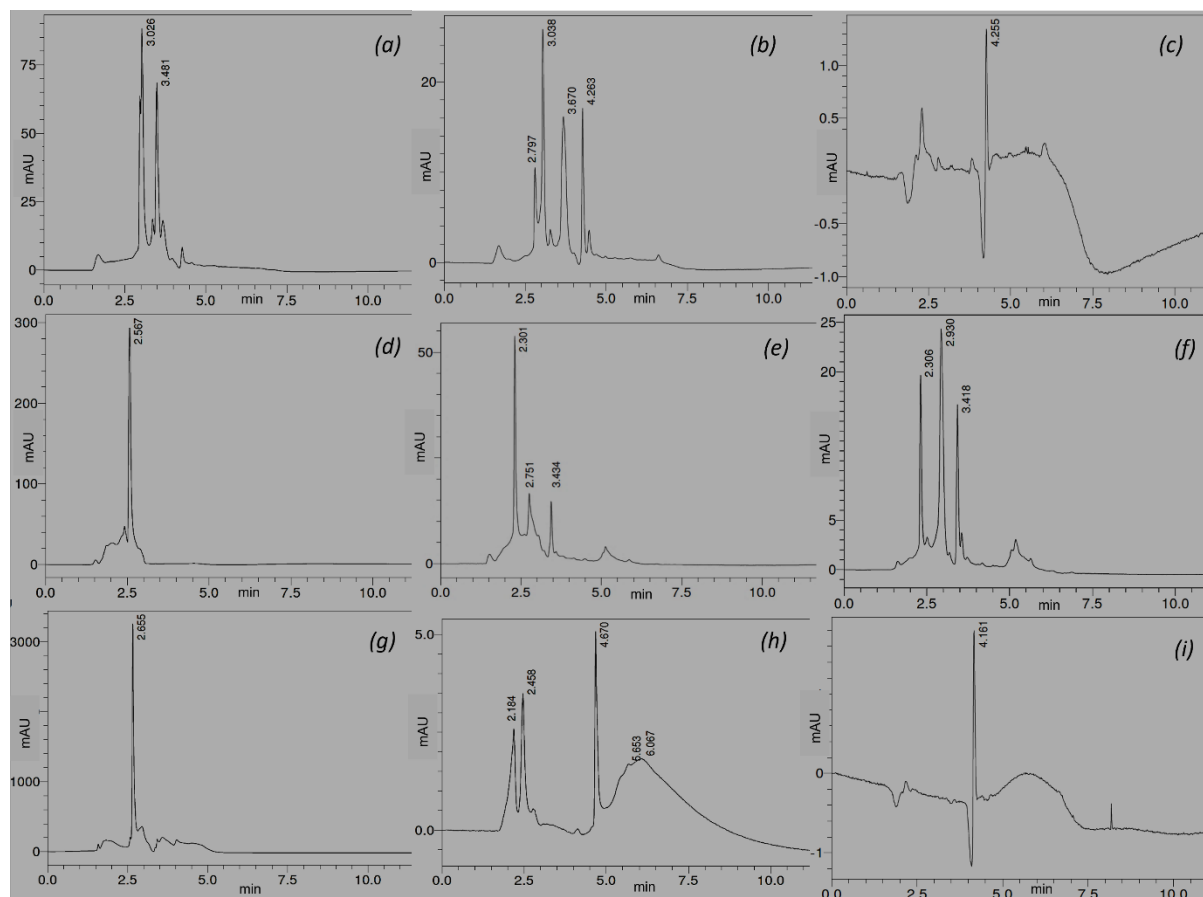
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471 **Fig. 3**

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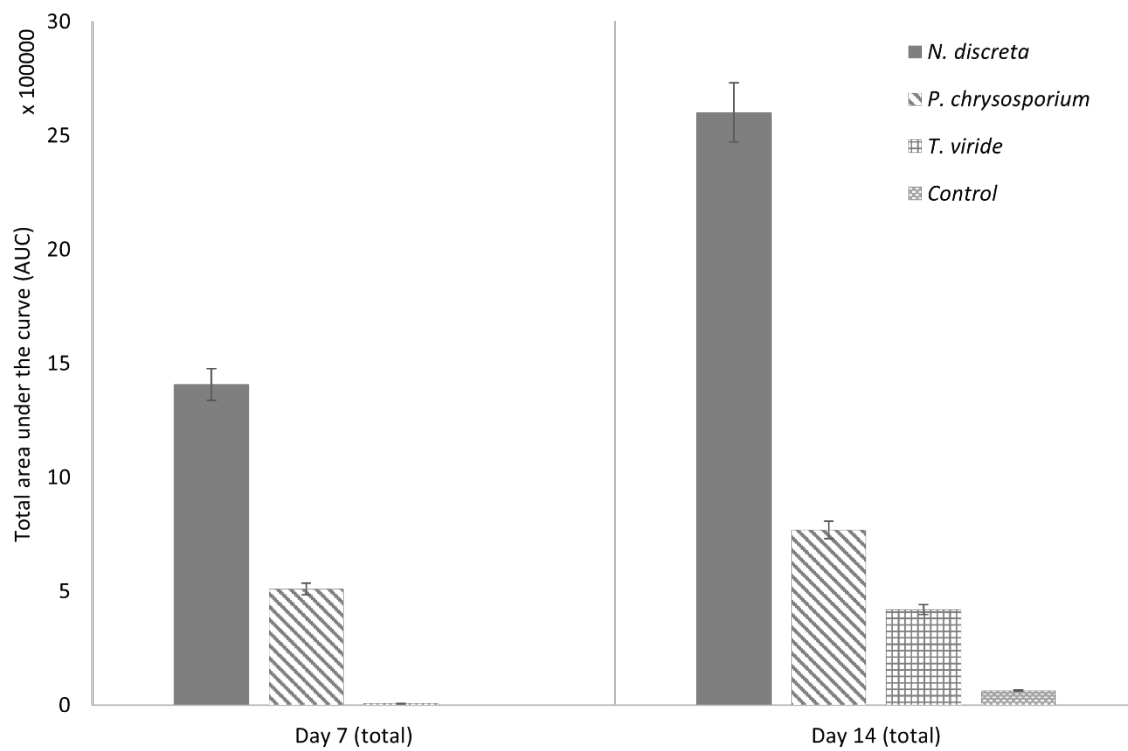
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476 **Fig. 4**

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